



Structure of ice surface : evidence for the existence of amorphous and non microporous ice



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Problem

- Measurement of high specific surface area (A)
- Measurement of low heat of adsorption (ΔQ): anomalously low if there is a pore confinement effect
- Does large specific surface prove microporosity ?
- Re-investigation of adsorption isotherm studies:
 - Comparison of amorphous (Ia) and crystalline (Ic) ices
 - Comparison of various adsorbates
 - Use of infrared co-measurements to check modifications in ice structure

Experimental conditions

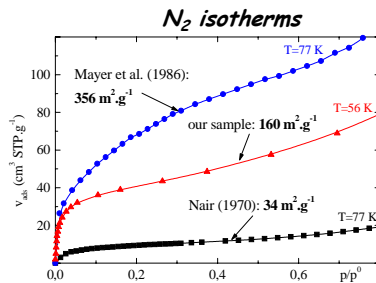
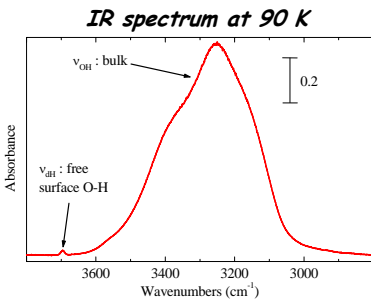
Ice formation

- H₂O:Ar (1:30) gas mixture sprayed into the cell at 40 K.
- Sample slowly annealed (0.2 K.min⁻¹) to 90 K at which ice is expected to be amorphous.

Adsorbates

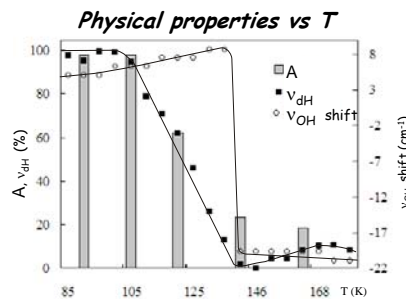
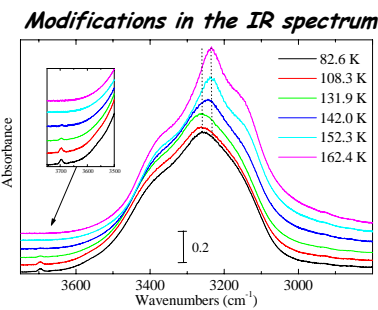
	N ₂	CO	CH ₄	Ar
Size (Å)	3-4.1	3.7-4.2	4.2	3.8

Ice sample characterization



- IR spectrum typical of that of amorphous ice
- Type II isotherms in agreement with other studies
 - Similar values of A : 100-300 m².g⁻¹
 - Similar values of ΔQ : 2.5 kJ.mol⁻¹
- Analogous physical properties for our samples and for those obtained directly by water vapor deposition

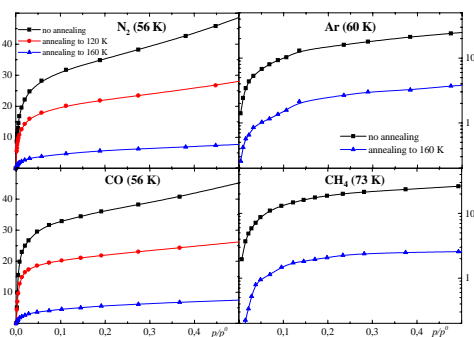
Ice sample annealing



- T>110 K : decrease in A and v_{dH}
- Surface re-arrangement before crystallization
- T>150 K : A(Ic) = 15 % A(Ia)
no more v_{dH} signal
- Less dH bonds for Ic than for Ia

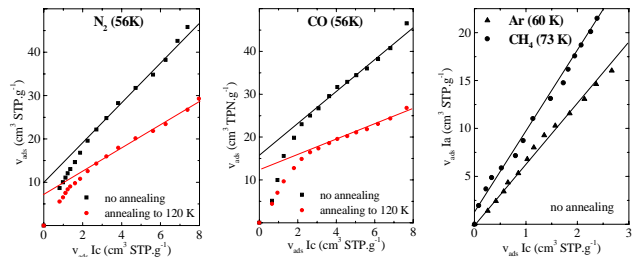
Microporosity analysis

Isotherms on Ia and Ic



- Crystalline ice is taken as reference of non porous material

α -plot



- Deviation from linearity observed only for CO and N₂
- Specific N₂-, CO-dH interactions rather than confinement effect

Conclusions

- N₂ is not suitable to probe porosity
- Evidence for the existence of amorphous and non microporous ice
- Model of grain assembly (size < 65nm) ?

- Open surface favours molecular mobility, diffusion and reactions
- Importance for the understanding of interstellar reaction mechanisms ?