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Methods :

experimental :

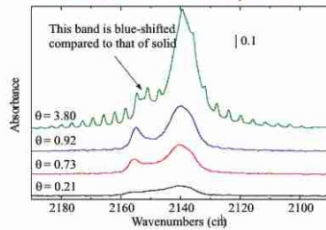
- combination of volumetric and infrared measurements to plot «IR isotherms» and to get additional information

theoretical :

- DFT calculations using Plane-Wave ultrasoft Pseudopotential to deal with large systems

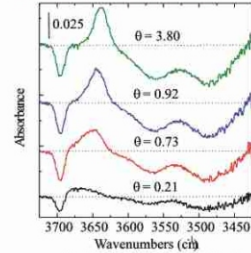
Experimental study

$\nu(\text{CO})$ infrared spectra



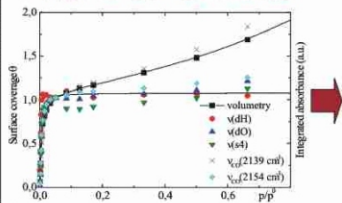
- 2 bands grow simultaneously
- blue shift \Rightarrow H bond

Surface modes of ice



- 3 interactions : dH, s4 and dO

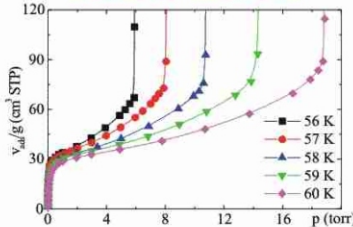
Plot of IR isotherms



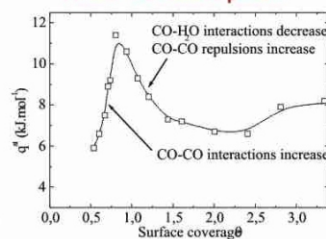
- For CO :
- ν_{CO} (2154 cm^{-1}) \Leftrightarrow CO above dH
 - ν_{CO} (2139 cm^{-1}) \Leftrightarrow monolayer + multilayer

- For ice :
- 3 surface modes with similar evolution

Series of isotherms



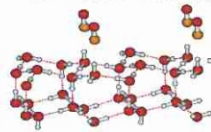
Heat of adsorption



Quantum study

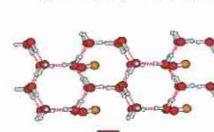
One CO per cell

- dH : 11.4 kJ/mol

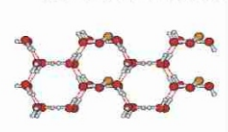


CO interacting with :

- s4 : 9.6 kJ/mol



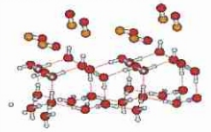
- dO : 10.4 kJ/mol



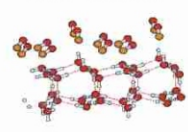
Low surface coverage : 3 equivalent adsorption sites

Higher surface coverage

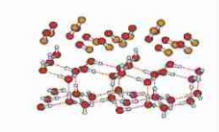
- 2 CO per cell (dH,s4): 13.3 kJ/mol



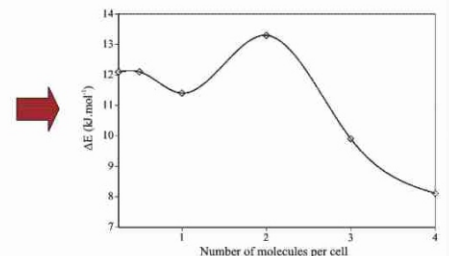
- 3 CO per cell: 9.9 kJ/mol



- 4th CO above the canal: 8.1 kJ/mol

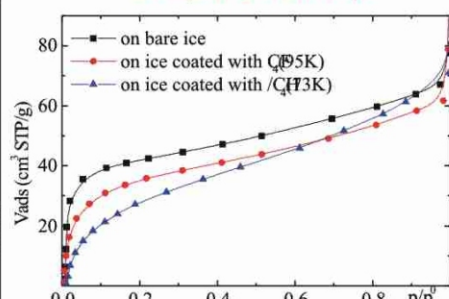


Adsorption energy depends on the number of molecules per cell



Successive adsorptions : CO/CF₄/Ice and CO/CH₄/Ice

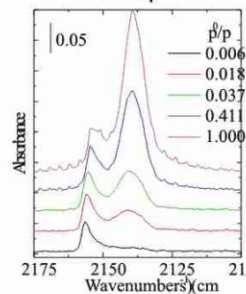
Volumetric isotherms



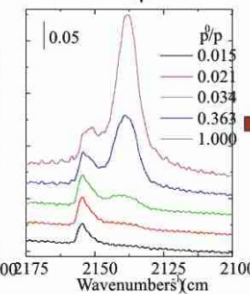
- CO adsorption disturbed by the pre-adsorbates
- Greater influence of CH₄ compared to CF₄

$\nu(\text{CO})$ infrared spectra

CO/CF₄/ice

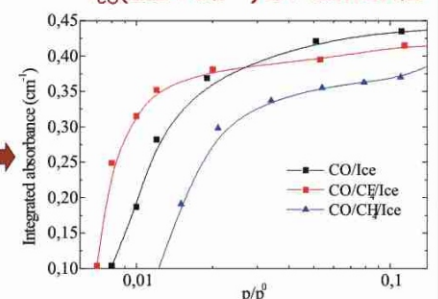


CO/CH₄/ice



- The 2 bands do not grow simultaneously: ν_{CO} (2154 cm^{-1}) grows alone at the beginning

ν_{CO} (2154 cm^{-1}) IR isotherms



- dH sites saturated at lower p/p⁰ when CF₄ is pre-adsorbed \Rightarrow CF₄ lets them free
- Weaker dH-CH₄ affinity than dH-CO

Conclusions

- Good agreement between experiment and theory
- Weak interaction between CO and ice
- At least 3 sorts of adsorption sites, but no evidence for site selectivity
- Lateral interactions are likely to stabilize
- Successive adsorptions reveal enhanced dH-CO affinity compared to dH-CF₄ and dH-CH₄